

# Minerals and Rocks

Chapter 20

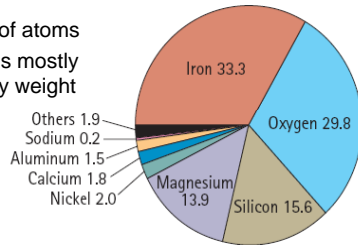


## Earth System Science

- Interconnected
- Rocks and minerals
- Interior processes
- Erosion and deposition
- Water and air

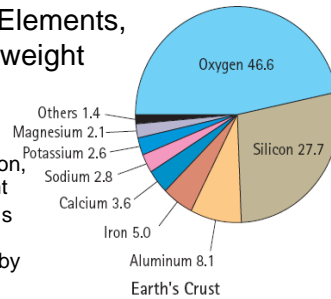
## Elements of Earth by weight

- Made of atoms
- Earth is mostly iron, by weight



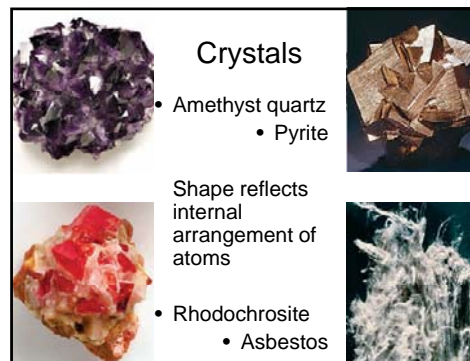
## Crust Elements, by weight

- Made of atoms
- Earth is mostly iron, by weight
- Surface is mostly oxygen, by weight



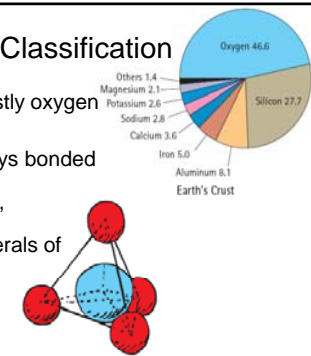
## Minerals

- Naturally occurring
- Not composed of 'organic' molecules
- Crystalline solid
- Specific chemical composition



## Mineral Classification

- Crust is mostly oxygen and silicon
- Silicon always bonded to oxygen
- 'SILICATES'
- 92% of minerals of crust



## Silicate Minerals

- Silica bonded to metals
- Aluminum, sodium, potassium, calcium
  - Feldspar: Most abundant mineral
  - 'felsic minerals'
  - Pale, average density
- Examples of felsic minerals
  - Feldspar
  - Quartz
  - Muscovite mica

## Silicate Minerals

- Silica bonded to metals
- Iron, magnesium
  - Ferromagnesian silicates: 'ferromags'
  - Dark, denser than felsic minerals
- Examples of ferromags
  - Amphibole
  - Pyroxene
  - Biotite mica
  - Olivine

Mineral	Mineral Formula	Cleavage	Silicate Structure
Olivine	$(Mg, Fe)_2SiO_4$	None	Single tetrahedron
Pyroxene group (Pyroxene)	$(Mg, Fe)SiO_3$	Two planes at right angles	Single chains
Amphibole group (Amphibole)	$C_2H_4(Mg, Fe, Al)_7Si_8O_{22}(OH)_2$	Two planes at 60° and 120°	Double chains
Biotite	$K(Mg, Fe)_3AlSi_3O_{10}(OH)_2$	One plane	Sheets
Muscovite	$KAl_3(AlSi_3)O_{10}(OH)_2$	One plane	Sheets
Orthopyroxene (Pyroxene)	$KMgSi_2O_6$	Two planes at 90°	Three-dimensional networks
Plagioclase	$(Ca, Na)Al_2Si_2O_8$	Two planes at 90°	Three-dimensional networks
Quartz	$SiO_2$	None	Three-dimensional networks

## Hardness

- Resistance to scratching
- Compare to glass/steel, penny, fingernail

## Breaking minerals

- Strength of bonds within crystals
- Cleavage
  - Some planes with weak bonding
  - Break along these



## Breaking minerals

- Strength of bonds within crystals
- Fracture
  - No planar arrangement of weak bonds
  - Conchoidal or irregular



## Non-silicates

- Carbonates
  - Calcite:  $CaCO_3$
- Oxides
  - $Fe_2O_3$ ,  $Fe_3O_4$
  - tin, chromium, uranium
- Sulfides
  - Zinc, lead, mercury
  - Pyrite:  $FeS_2$
- Native elements: Au, Cu

## Minerals crystallize

- From liquid (usually) or gas (occasionally)
- Magma: molten rock
- Watery solutions

## Crystallization of Magma

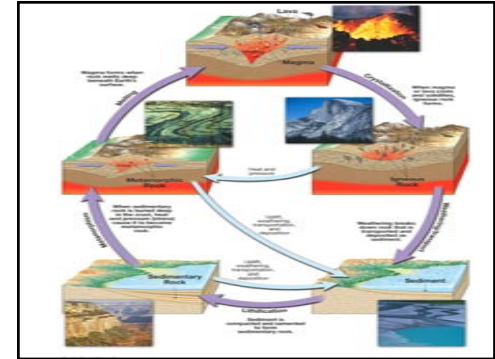
- Cools, atoms attracted to one another
- Arrange in orderly crystalline structures
- When very hot, low-silica forms
- Cooler, greater amounts of silica in them
- Composition of magma changes as crystallization proceeds

## Crystallize from watery solutions

- Change solubility by changing physical or chemical conditions in magmatic water left
  - pH, other ion content
  - Temperature, pressure
- Chemical sedimentary rock
  - Carbonates: made by organisms, mostly
  - Increase concentration by evaporation: evaporites

## Rock Types

- Igneous
- Sedimentary
- Metamorphic



## Sedimentary rocks

- Cover 2/3 of Earth's surface
- Record conditions at time of deposition
- Include remains of organisms preserved as fossils

## Sedimentary rocks

Sediment is derived from weathering  
Carried by fluid  
Formed at Earth's surface  
Important to reconstruct much of Earth's history

## Sedimentary rocks

### Features of sedimentary rocks

- Strata, or beds (most characteristic)
- Bedding planes separate strata
  - May have important characteristics
- Size, shape and distribution of grain sizes
- Fossils

## Sedimentary rocks

Two main types

- Rocks formed by deposition of sediment—**Clastic**
- Rocks formed by precipitation from water--**Chemical** (includes rocks formed by organisms)

## Clastic Sediment Grains

- Particle loosened from pre-existing rock
- Transported to place of deposition
- Shape, size, and sorting of grains can tell about the environment of deposition

## Lithification

### Process of becoming stone

- Burial and compaction
- Precipitation of cement
- Each reduces 'pore space'

## Cement

- Brought in by water
- Mineral material between grains
- Fills in pore spaces
- Commonly calcite, silica, and sometimes iron oxide

## Bedding and bedding planes



• [http://www.birdandhike.com/Hike/General\\_Info/Glossary/Gloss4.htm](http://www.birdandhike.com/Hike/General_Info/Glossary/Gloss4.htm)

## Types of Clastic Rocks

- Shale (most abundant)
- Sandstone
- Conglomerate

## Fossils

- Traces or remains of prehistoric life
- Are the most important inclusions
- Help determine past environments
- Used as time indicators
- Used for matching rocks from different places

## Shale with plant fossils



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## Shale

- Composed of very fine grained sediment
- Shows obvious tendency to split along planes (fissile)
- Usually gray
- Most common type of sedimentary outcrop

## Sandstone



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## Sandstone

- Composed of sand-size particles
  - Between 1/16 mm and 2 mm diameter
  - Particles may be individual mineral grains or rock fragments
  - Quartz most common type of grain
- Environments include
  - Beach,
  - river,
  - shallow sea,
  - sand dunes

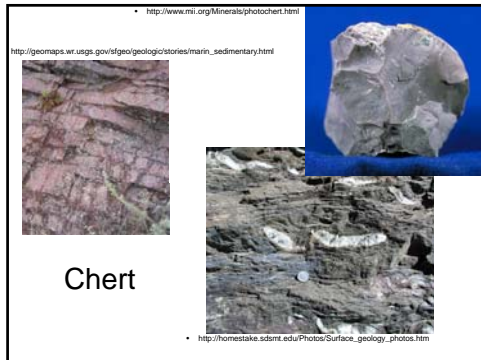
## Conglomerate



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### Classification of sedimentary rocks

Detrital Sedimentary Rocks			Chemical Sedimentary Rocks		
Sediment (grain size)	Sediment Name	Rock Name	Composition	Texture (grain size)	Rock Name
Coarse (bigger than 2 mm)	Gravel (Rounded fragments) Gravel (Angular fragments)	Conglomerate Breccia		Fine to coarse crystalline	Crystalline Limestone Travertine
Medium (1/16 to 2 mm)	Sand (If abundant fossils, it is called fossiliferous)	Sandstone	Calcite, CaCO <sub>3</sub>	Medium, shaly, and shaly fragments loosely cemented	Conglomerate Limestone
Fine (1/16 to 1/250 mm)	Mud	Siltstone		Thin, shaly, and shaly fragments cemented with calcite cement	Fossiliferous Limestone
Very fine (bigger than 1/250 mm)	Mud	Shale		Microscopic shells and clay	Chalk
			Quartz, SiO <sub>2</sub>	Very fine crystalline	Chert (light colored) Flint (dark colored)
			Common CaCO <sub>3</sub> , MgCO <sub>3</sub>	Fine to coarse crystalline	Rock Opalium
			Halite, NaCl	Fine to coarse crystalline	Rock Salt
			Altered plant fragments	Fine-grained organic matter	Bituminous Coal

- ### Features of sedimentary rocks
- Porosity
  - Permeability

- ### Sedimentary rocks
- #### Economic importance
- Coal
  - Petroleum and natural gas
  - Precipitation of iron and aluminum
  - Deposition of gold and tin
  - Sand, gravel, clay